

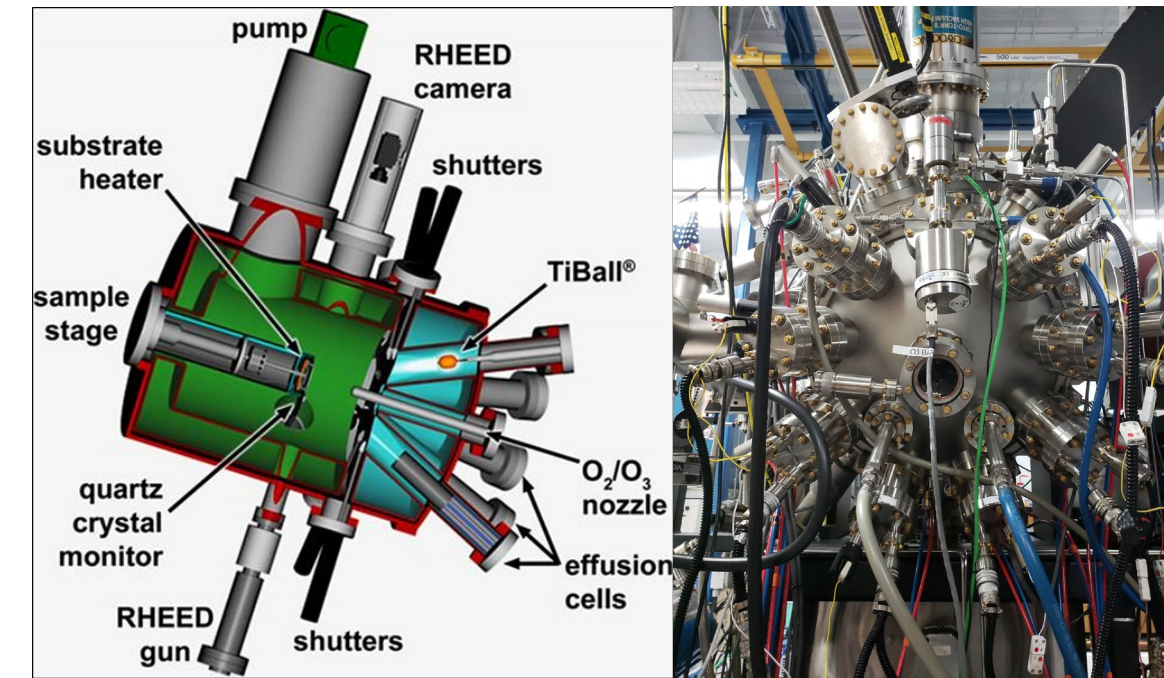
# Electronic and magnetic order as a function of doping in mixed-valent $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$ thin films

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## Molecular Beam Epitaxy

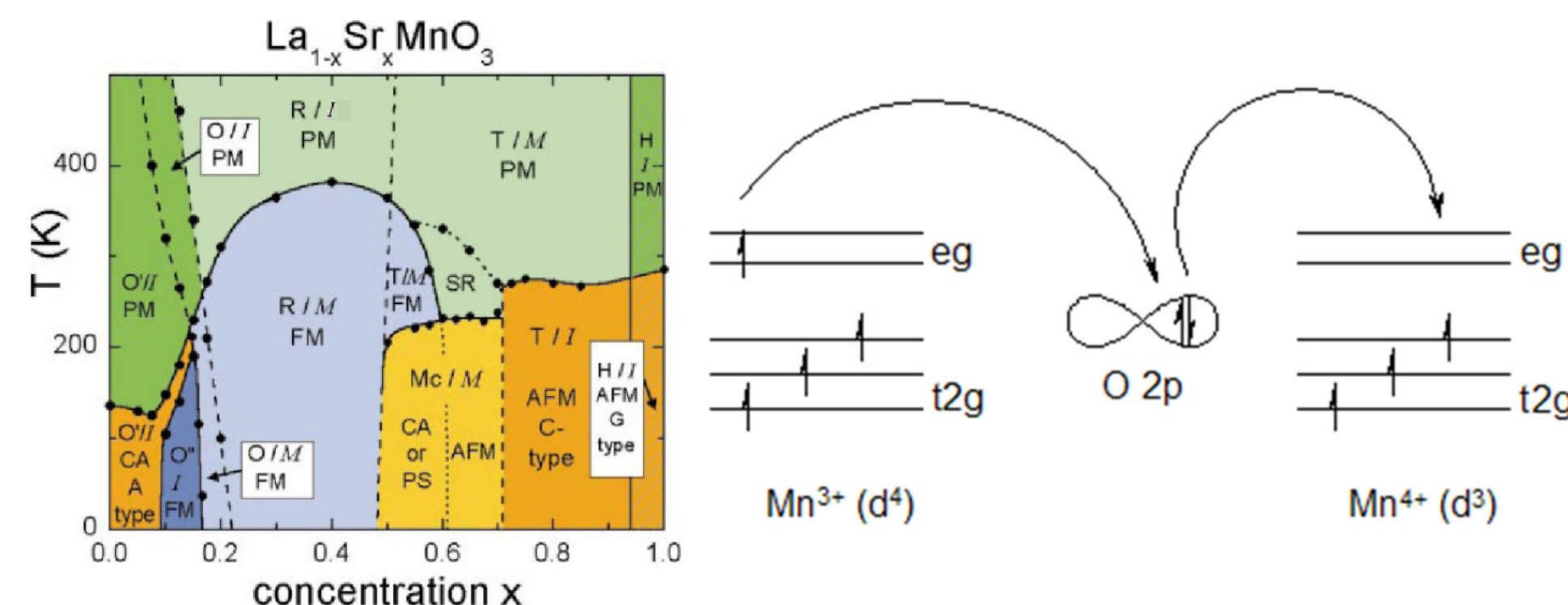
- Our samples were grown using a technique known as Molecular Beam Epitaxy (MBE).
- In MBE, we utilize carefully calibrated molecular beams of constituent elements to produce thin films one atomic layer at a time
- Sources of the constituent elements are heated in effusion cells to evaporate material into the chamber, forming molecular beams.
- In oxide MBE, a form of oxygen is introduced into the chamber to form oxides.
- For our growths of  $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$ , ozone was employed as our source of oxygen, allowing us to keep the chamber pressure low ( $5\text{E-}7$  Torr) due to ozone's high oxidizing potential.



**Fig. 1** (Left) A side view diagram of an MBE chamber. (Right) A front view picture of the chamber used in our growths.

## Bulk Phase Behavior

- The behavior of  $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$  has been extensively documented in bulk crystals with respect to temperature and strontium concentration.
- Undoped  $\text{LaMnO}_3$  is known to be a Mott insulator that at higher temperatures has paramagnetic behavior and becomes antiferromagnetic at lower temperatures.
- In the range around  $x=0.4$   $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$  is a paramagnetic insulator at higher temperatures but at lower temperatures it transitions to a ferromagnetic conductor.
- This ferromagnetic-conducting state is kinetically favorable around  $x=0.4$  at lower temperatures because the double exchange interaction allows for the delocalization of electrons across spin aligned manganese atoms, reducing free energy.

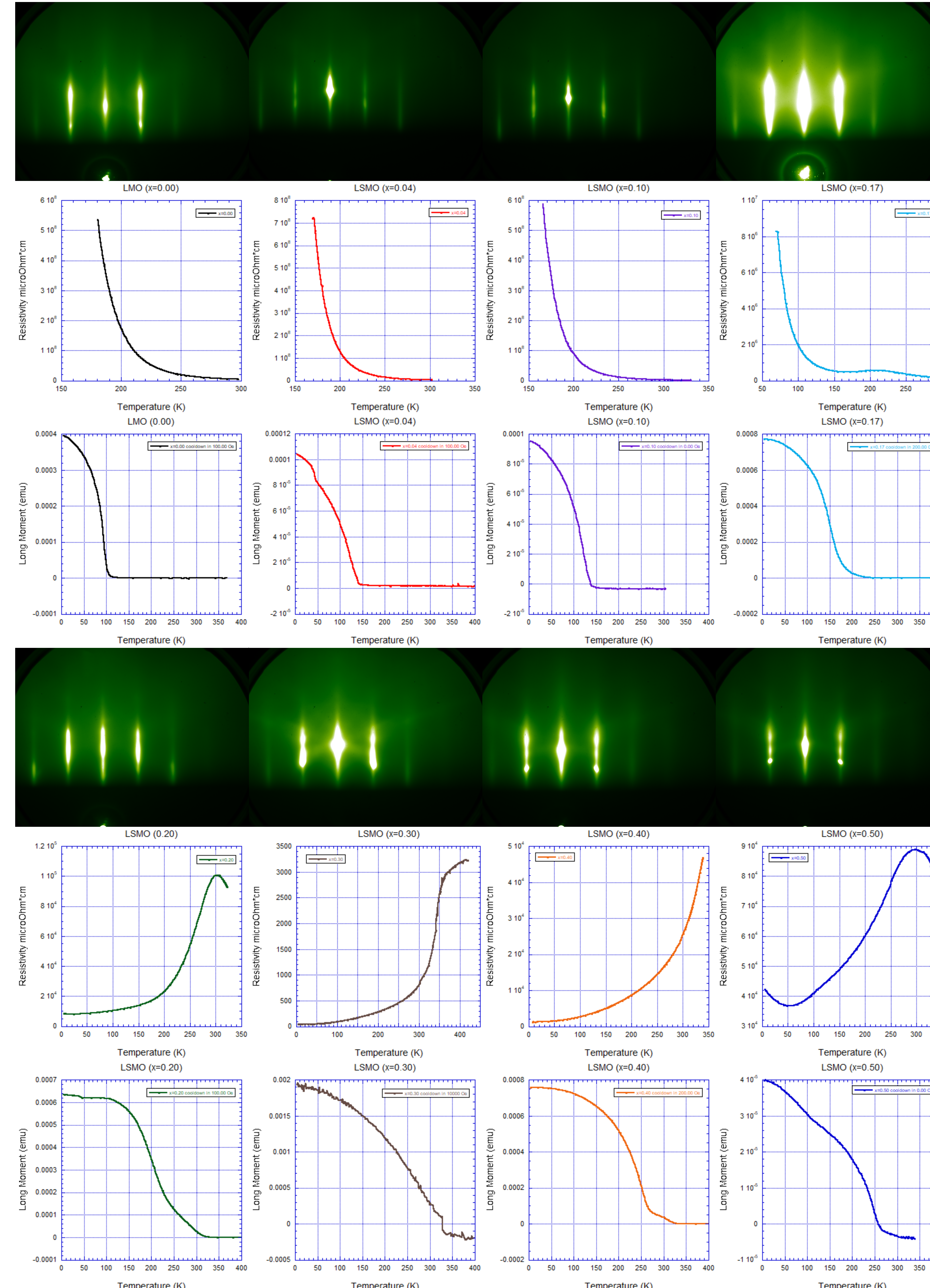


**Fig. 3** (Top) The phase diagram of  $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$  in bulk.<sup>1</sup> (Bottom) Mechanism for the double exchange interaction.

## Reference

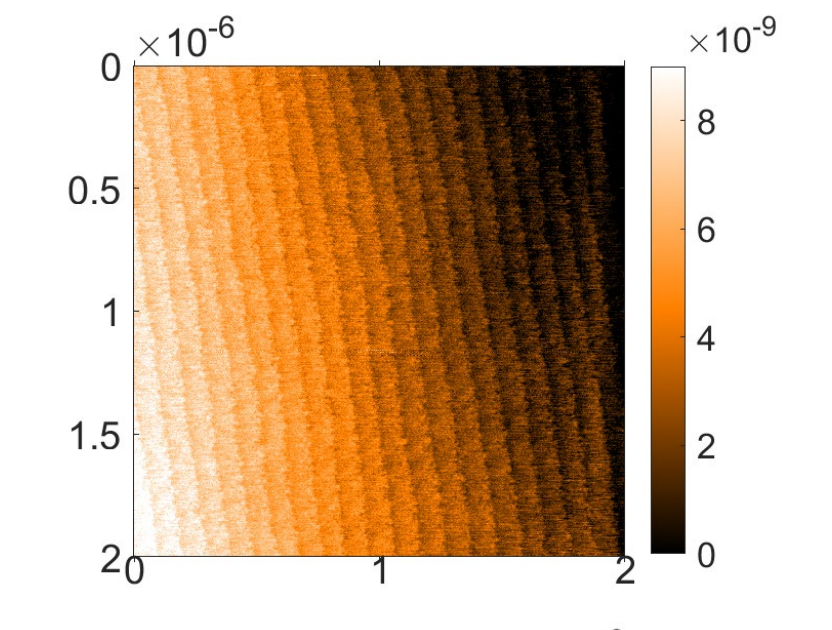
[1]Hemberger *et al*, PRB 66, 094410 (2002).

## Thin Films of $\text{La}_{1-x}\text{Sr}_x\text{MnO}_3$ for Different Doping Concentrations

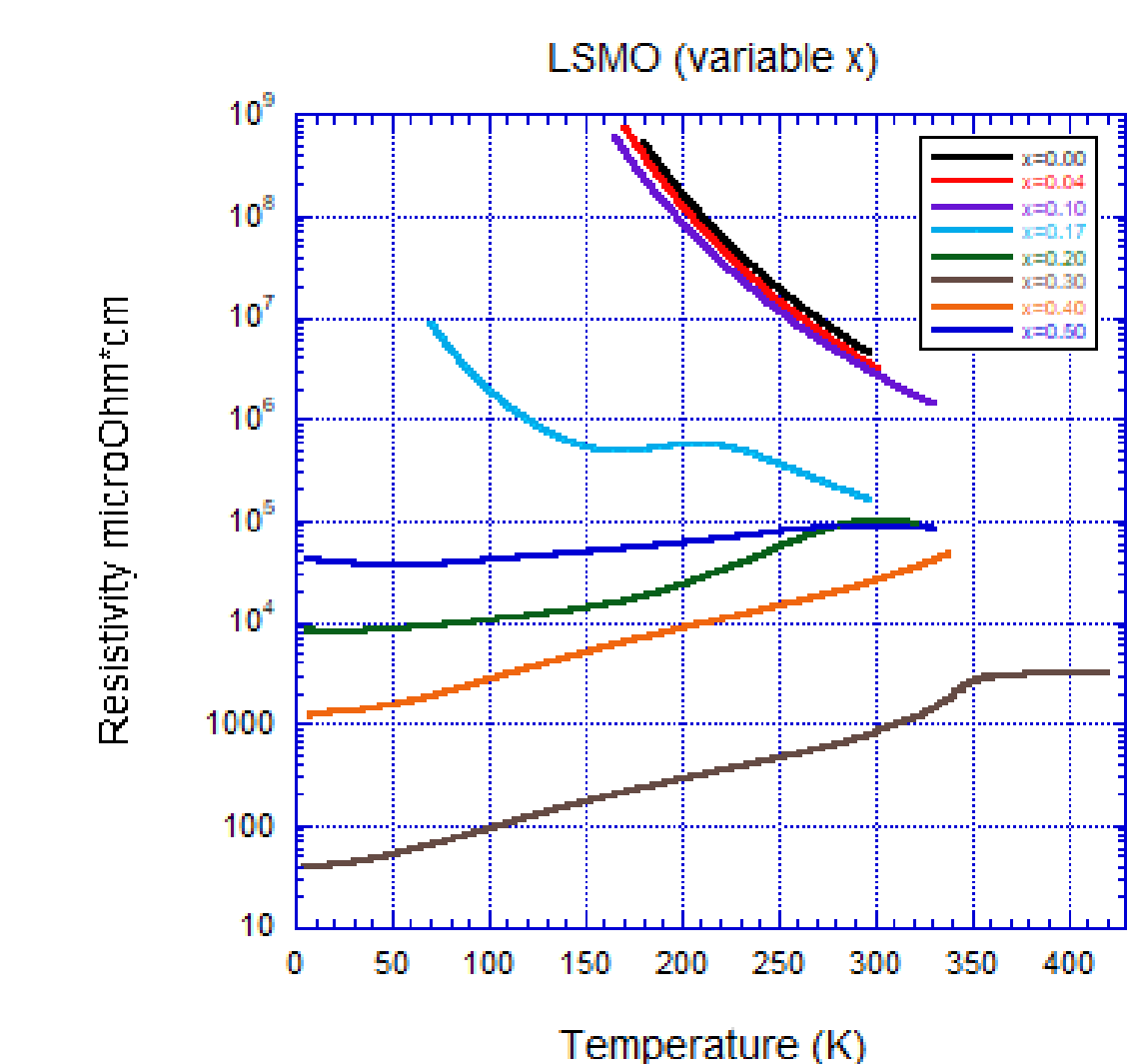


**Fig. 5** (Top) Reflection high energy electron diffraction (RHEED) patterns taken during the growth of each sample ( $x=0, 0.04, 0.10, 0.17, 0.20, 0.30, 0.40, 0.50$ ). (Middle) The measure of film resistivity as a function of temperature for each sample. (Bottom) The measure of film magnetic field, collected in a field cooldown, as a function of temperature for each sample.

- An atomic force microscopy scan was done for the  $x=0.30$  film. In this scan, figure 4, terraces can be seen in the surface morphology, which is indicative that the film is flat and strained to the substrate.
- Additionally, Reflection high energy electron diffraction (RHEED) patterns taken near the end of the growths, shown in figure 5, also indicate well-strained growth and a 2-dimensional surface morphology.
- From the film transport and magnetic data, figure 5 and 6, we can observe transitions in the electronic and magnetic ordering of the samples.
- Samples  $x=0, 0.04$ , and  $0.10$  are shown to be insulators over the entire temperature range measured, with magnetic ordering in the  $x=0$  sample below around  $100\text{K}$  and below  $140\text{K}$  for the  $x=0.04$  and  $0.10$  samples.
- In  $x=0.20, 0.30$ , and  $0.40$  we observe a ground state conducting behavior, while insulating behavior is present below  $50\text{K}$  in  $x=0.50$  and  $170\text{K}$  in  $x=0.17$
- A metal-to-insulator transition accompanied by magnetic ordering is observed below around:  $210\text{K}$  for  $x=0.17$ ,  $310\text{K}$  for  $x=0.20$ ,  $400\text{K}$  for  $x=0.30$ , and  $260\text{K}$  for  $x=0.50$ . We also observe magnetic ordering in  $x=0.4$  below around  $320\text{K}$  though no metal-to-insulator transition is observed within the measured range.



**Fig. 4** Atomic force microscopy (AFM) image of the surface of the  $x=0.3$  sample.



**Fig. 6** Film resistivity as a function of temperature for  $x=0, 0.04, 0.10, 0.17, 0.20, 0.30, 0.40$  and  $0.50$  compared on a log scale.